

Nitrous oxide and carbon dioxide: their similar and contrasting biological effects

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The similar and contrasting biological properties of CO₂ and N₂O appear to reflect fundamental biological processes in both very simple and the most complex organisms.

Introduction

A recent article in this journal¹ discussed the similarities and differences between nitrous oxide (N₂O) and carbon dioxide (CO₂) from a physico-chemical perspective. Here we consider the apparent similarities and differences between these two gases in a biological context.

It is well known that biologically active chemical agents can act directly with receptors in the brain to produce physiological and pharmacological effects.² These receptors have unique and specific physico-chemical structures and properties. Chemical substances that attach to these receptors do so by characteristic physico-chemical mechanisms, which then trigger a cascade of events through the interaction of the particular agent with the specific receptor.² It is therefore not surprising that the structural similarities of the N₂O and CO₂ molecules translate into similar biological actions. On the other hand, such structural and other differences as they have may account for their dissimilar biological effects.

Nitrogen and carbon cycles

Photorespiration involves the light-dependent evolution of CO₂ and uptake of oxygen in green leaves. During the 1950s and 1960s, the synthesis of glycolate by leaves in high O₂ concentrations or low CO₂ settings was demonstrated and the enzymic pathways involved were elucidated.³ Ribulose biphosphate carboxylase/oxygenase (rubisco, RuBP) is the primary enzyme involved, with both CO₂ and O₂ competing for its active site, the position on the enzyme to which the substrate binds. There is a well-established link between photorespiration and the nitrogen cycle in plants. During the decarboxylation of glycine, ammonia is produced in stoichiometric amounts with

the CO₂ involved in photorespiration. Plants acquire their nitrogen either directly from the soil or through internal recycling of nitrate and ammonium. The amount of ammonium recycling in C₃ plants is significantly higher than in C₄ plants as a result of photorespiration. As the rate of ammonia production in C₃ plants can be 20 times the rate of primary nitrate assimilation, the plant cannot afford to lose this ammonia as a volatile gas but must have an effective mechanism for its re-assimilation. Three enzymes (glutamine synthetase, glutamate synthase and glutamate dehydrogenase) have been identified in transforming ammonia into amino acids but there is debate as to the role of the various oxidized nitrogen intermediates which are formed.³ It is possible that one of these oxides is NO, which may be reduced to N₂O as has been shown in mammalian cells⁴ and bacteria.⁵ In support of this idea, plants have been shown to emit endogenously produced N₂O during nitrate assimilation.⁶

Nitrous oxide does not substitute for O₂ in either the cytochrome *c* oxidase or carbon monoxide (CO) diogenase reactions that produce, respectively, water and CO₂ and which are both catalysed by cytochrome *c* oxidase.⁷ Rather, it acts by reducing the rate of electron transfer from cytochrome *c* to the O₂ reaction site.⁷ It is therefore possible that N₂O interacts with CO₂ in the plant cell by reducing the flow of electrons to both CO₂ and O₂ reaction sites. The latter postulate is supported by the finding that N₂O inhibits the use of O₂ by plant tissue.⁸

The link between N₂O and CO₂ through the carbon and nitrogen cycles is also seen in the composition of the atmosphere and oceans during glacial cycles.⁹ The end of a glacial cycle is marked by the outgassing of N₂O, a greenhouse gas, from sea water. This is described by a nonlinear model which takes into account the effect of variations in the ocean's content of fixed nitrogen following enhanced denitrification.⁹ Outgassing leads to greater levels of atmospheric CO₂ during interglacial periods,⁹ showing a direct relationship

between CO₂ and N₂O in the atmosphere and ocean.

Nitrous oxide, like carbon dioxide, is produced endogenously

Nitrous oxide is a metabolic product of certain bacteria, some of which are commensals in the gut of animals including man.¹⁰ In recent years, moreover, evidence has appeared indicating that N₂O may be produced in mammalian cells as part of their metabolism.^{4,7} When we discovered that N₂O could be involved in neurotransmission in humans,^{11–13} we did not realize that we had stumbled on a new biological principle. The generality of our finding was confirmed when nitric oxide (NO), another endogenously produced gas, was also discovered to have a role in neurotransmission.^{14,15}

The evidence that N₂O is produced endogenously^{4,7} seems to confirm the possibility that it is not only the first gas to have been shown to have a role in neurotransmission but that it is formed by the catabolism of NO.^{16,17} This once again highlights the similarities of N₂O and CO₂, which both appear to be endogenously produced as metabolic products in bacterial and mammalian cells. These gases are therefore involved in the physiology of cells from the simplest to the most complex life forms. This is perhaps not surprising, given that N₂O and CO₂ are linked through the nitrogen and carbon cycles (see above), which are crucial to living organisms.

Actions on the nervous system

In higher animals, N₂O and CO₂ have related actions on the nervous system. Both gases in high concentrations cause depression of the central nervous system, leading eventually to coma, respiratory depression and finally death.^{18–20} Despite these apparent similarities, however, there are differences in their biological effects.^{19,20}

Respiration

The effects of N₂O on respiratory drive are relatively small, with slight depression of the stimulatory response in humans to CO₂ when 50% N₂O in oxygen is breathed.¹⁹ The response to hypoxia is also reduced when 50% N₂O is breathed. Since N₂O at non-anaesthetic concentrations (such as 50%) is a multipotent opioid agonist,^{21–23} its depressant effect on respiration could be opioid mediated. Nevertheless, it is much less marked than some other opioids and particularly those that are selective mu-opioid agonists, such as morphine. Competitive opioid antagonists such as

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nalorphine and naloxone can reverse these effects on respiration.^{24,25} Such antagonistic outcomes on respiratory depression are extremely rapid as would be expected because they are mediated by direct effects on opioid receptors. Moreover, endogenous and exogenous opioids, including N₂O, inhibit the responses of the brainstem respiratory centre to increased CO₂ blood levels.¹⁹ Thus, the control of respiratory function 'may be directly related to competition for opioid receptor sites by these various gases as well as endogenous opioid agents.'²² It is therefore significant that the depressant effect of morphine on respiration can be as rapidly reversed by inhalation of 5% CO₂ in oxygen,²⁶ indicating the possibility that this effect is also receptor mediated.

That this influence of CO₂ is likely to involve receptors is supported by the fact that stimulation of ventilation in man begins within seconds of inhaling low concentrations of the gas, with a maximum response occurring within less than 5 minutes.²⁰ These stimulatory consequences of CO₂ inhalation disappear rapidly after its withdrawal.²⁰ Indeed, the CO₂ effect on medullary chemoreceptors and peripheral arterial chemoreceptors are clearly receptor mediated.²⁰ In these circumstances, it is likely that the effects of CO₂ on the ventilatory neurons in the medullary-pontine region are also mediated by receptors. A further clue that this is an opioid interaction is that, in barbiturate poisoning, which is not mediated by opioid receptors,²⁷ the administration of CO₂ has no positive effect on stimulating respiration.²⁶ That various gases influence respiration by interacting directly at opioid receptors is supported by the finding that 100% oxygen altered the *in vitro* binding of ³H-dihydromorphine as well as that of ³H-ketocyclazocine.²⁸ Dihydromorphine and ketocyclazocine are opioid receptor ligands, so that evidence that *in vitro* binding with these agents is interfered with by oxygen indicates that oxygen itself also binds to opioid receptors and therefore has opioid properties. Other studies have confirmed this work.³¹

Although binding studies with CO₂ using opioid ligands have not been performed, as far as we are aware, the similarity of the actions of the specific opioid antagonistic, naloxone, on CO₂ and N₂O analgesia (see below)²⁹ may indicate that CO₂ also binds to opioid receptors, to exert some of its respiratory effects.

The pharmacological effects of N₂O are also rapid in onset and offset^{19,30} and these

consequences are almost certainly receptor mediated^{16,17,21,22,28,31-33} and could also be responsible for its actions on the respiratory centres.

Sympathetic activity

Sympathetic activity, which prepares the body for the fright/flight response, is increased by both N₂O^{34,35} and CO₂.²⁰ Nitrous oxide seems to act directly at opioid receptors as well as via the alpha₂-adrenergic system.³⁶ Carbon dioxide potentially activates the central adrenergic system²⁰ and there is some preliminary evidence that it acts directly on mu-opioid receptors,²⁸ further underlining the similarity between the biological actions of these two gases. The descending adrenergic and nor-adrenergic cell groups form most of the lateral medullary reticular formation and are linked to various vital centres in the medulla oblongata of the brainstem, which includes the respiratory centres.³⁷

Apart from the influence of these gases on the adrenergic system, they also have local effects. One which they share is the direct local influence on systemic blood vessels, resulting in vasodilatation.^{20,38,39} The results of central sympathetic activation are usually opposite to the local effects of CO₂, whereas the local influence on blood vessels seems to be more important than are the adrenergically mediated vasoconstrictor effects.²⁰ Nitrous oxide^{35,36,38} shares these properties.

Pain perception

Analgesia can be produced by hypercapnia, which triggers the release of endogenous opioids, presumably from the spinal dorsal horn,²⁹ a region in the spinal cord involved with pain perception. The dorsal horn appears to be implicated because the modification of tail-flick latency persists after spinalization.²⁹ The authors responsible for these results concluded that hypercapnia may produce direct excitatory effects on spinal or bulbo-spinal enkephalinergic (that is, opioid) interneurons. Since this effect was also present in animals spinalized at lower thoracic levels, at least one population of these neurones must have been in the lumbo-sacral region.²⁹ Gamble and Milne draw the analogy with N₂O when they state that: 'It is interesting that analgesia induced by the structurally similar compound nitrous oxide can be partly blocked by relatively high doses of naloxone under some circumstances.' They found that analgesia caused by CO₂ was also blocked with relatively high doses of naloxone (2 mg/kg).²⁹

Other workers have shown that both N₂O and hyperventilation have similar inhibitory effects on lamina V of the spinal cord, where there is convergence of visceral and high threshold cutaneous pain.⁴⁰ These laminae are rich in opioid terminals.⁴¹

These findings seem to indicate that CO₂, like N₂O, acts directly on mu-opioid receptors, especially in their effects on pain perception and possibly also in their actions on respiration as suggested previously (see above) on the basis of radio-receptor assays.²²

Other pharmacological actions of CO₂ and N₂O on the central nervous system

Pharmacological, rather than physiological, concentrations of CO₂ produce stress reactions such as anxiety and fear.^{42,43} This outcome is probably mediated by stimulation of the sympathetic system²⁰ and blockade of opioid receptors.²⁶ In contrast, although also heightening sympathetic stimulation,^{19,20} non-anaesthetic doses of N₂O have anti-stress effects⁴⁴⁻⁴⁷ where the agonistic influence of N₂O on the opioid receptors seems to outweigh its sympathetic activity.⁴⁵ Indeed, low doses of N₂O reduce cortisol release.⁴⁶

Both CO₂ and N₂O have antidepressant effects.⁴⁸⁻⁵⁰ Interestingly, Silver found that such effects of CO₂ were magnified if preceded by a short administration of anaesthetic concentrations of N₂O for 8 or 9 inhalations. Nitrous oxide at 100% used as an anaesthetic, also for a few inhalations (less than a minute's exposure), has an antidepressant action.⁴⁹ Furthermore, inhalation of 80% N₂O (with 20% O₂) for between 1.5 and 5 minutes⁵⁰ demonstrated that the gas has antidepressant effects. Low concentrations of N₂O mixed with high concentrations of O₂ titrated to a level where the patient is relaxed but fully conscious [so-called psychotropic analgesic nitrous oxide (PAN)]¹⁷ also have marked antidepressant properties.⁵¹ Moreover, this work^{50,51} indicates that endogenous depressions (which are caused by an underlying biological disturbance) are resistant to the effects of N₂O, whereas reactive depressions (which are related to a stressful life event) are ameliorated by exposure to the gas.

Carbon dioxide at high concentrations (75% in O₂ administered for 30-120 seconds) is effective for treating addicts from whom abused substances such as alcohol^{52,53} and opioids^{54,55} are suddenly withdrawn.⁵⁶ PAN has also been used successfully to treat both acute alcohol^{53,57} and opioid^{58,59} withdrawal states.

Conclusions

Although we have touched on only some of the properties of these gases, the similar but distinct differences in molecular structure of CO₂ and N₂O seem to be reflected in their similar but contrasting biological actions. It is clear that these gases have overlapping effects. However, CO₂ stimulates respiration whereas N₂O has a slight antagonistic influence on the stimulatory consequences of CO₂. Another distinction is that pharmacological concentrations of CO₂ produce a stress reaction, while the reverse is true of N₂O. Moreover, although both gases can ameliorate withdrawal states after substance abuse, the duration of exposure and the concentrations of the gases required are different.

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